

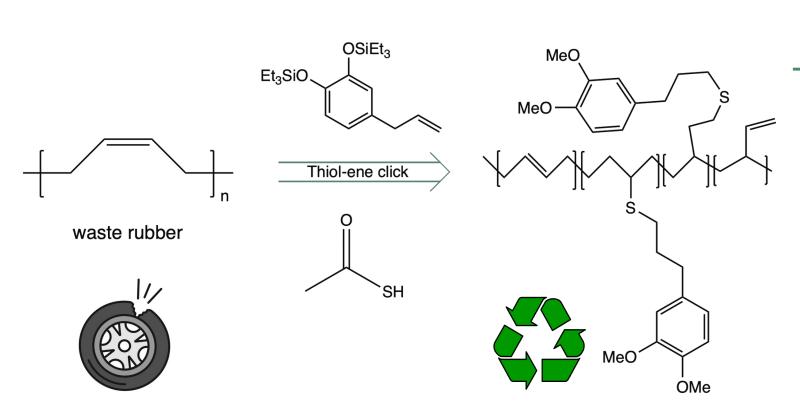
Upcycling of Rubber into Adhesives

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Abstract



The synthesis of a polycatechol was carried out by subjecting polybutadiene (PB) to a thiol-ene click reaction to reduce end of life tire waste.

Synthetic rubbers are not amenable to modern recycling methods, and the tire industry is the largest market segment. Polymer upcycling, which uses polymer waste to create new products of higher value, provides a promising incentive to recycle these discarded tires and give them new purposes. Our research transformed PB into a polycatechol product with adhesive properties greater than scotch tape. Polycatechols derived from PB simultaneously reduce environmental pollution and produce value-added products from the abundant feedstock of end-of-life tires.

Justification & Motivation

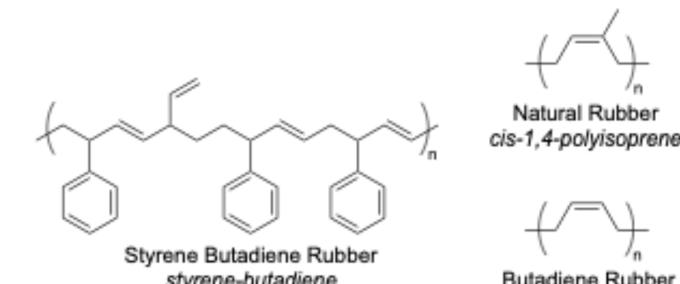
Importance of Recycling Polybutadiene



4 billion tires sit in landfills around the world. 1-2

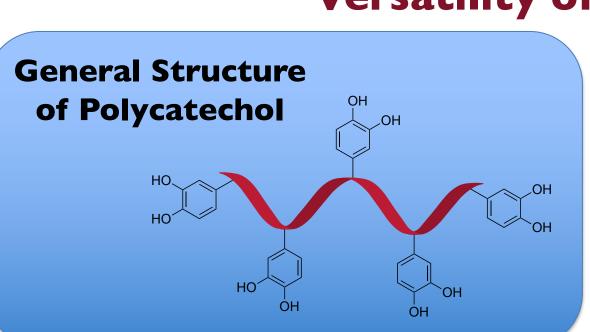
Myocardial Tissue Repair Hepatic Tissue Repair

Self-healable adhesive



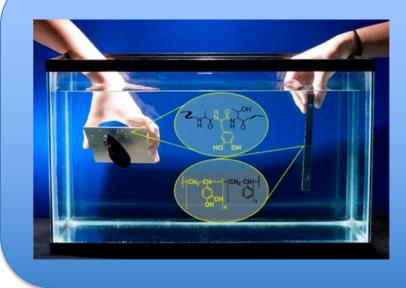
Types of rubbers used in tires.³

Versatility of Polycatechol

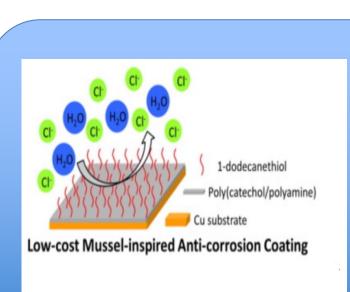


High wet adhesion strength

*** = EPL *** = PEG







Anti-Corrosion Coatings Poly(catechol/ amine) anticorrosion coating on copper

substrate.6

polybutadiene

Material

Adhesives

Mussel mimic

adhesive that

underwater.4

can bind

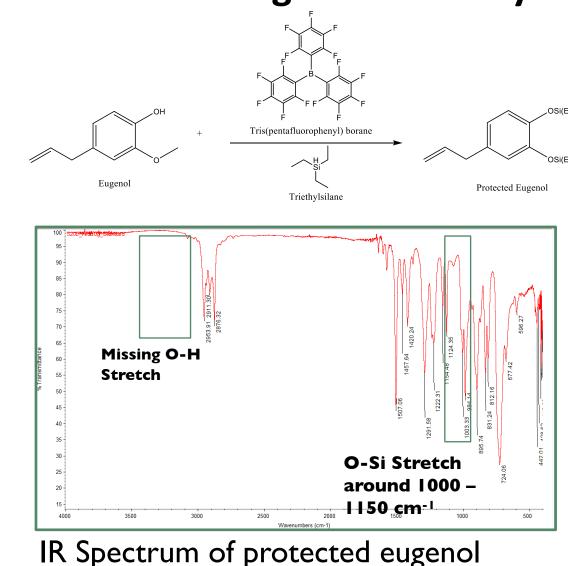
- Agency, A. Environmental disaster: World's biggest tire graveyard in Kuwait. Daily Sabah. Aug 13, 2021. (accessed 2024-04-23). 2. Scott, A. Can Tires Turn Green? Tire Manufacturers Are Adopting Greener Production Processes and More Renewable Materials, but They Have yet to Get a Grip on Tire Particle Pollution. Chemical & Engineering News. May 26, 2023. (accessed 2024-04-23). 3. Fazli, A.; Rodrigue, D. Recycling Waste Tires into Ground Tire Rubber (GTR)/Rubber Compounds: A Review. J. Compos. Sci. 2020, 4(3), 103. DOI:
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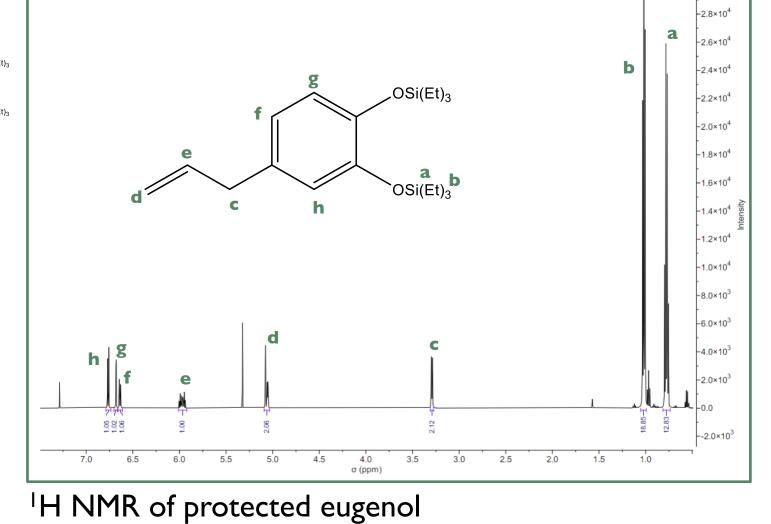
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- 6. Wu, J.; Cai, C.; Zhou, Z.; Qian, H.; Zha, F.; Guo, J.; Feng, B.; He, T.; Zhao, N.; Xu, J. Low-cost mussel inspired poly(catechol/polyamine) coating with superior anti-corrosion capability on copper. Journal of Colloid and Interface Science 2016, 463, 214-221. DOI: 10.1016/j.jcis.2015.10.056

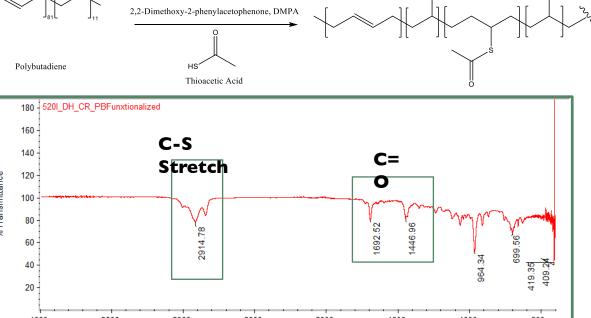
Synthesis of Catechol & Thioacetate Polybutadiene

Protection of Eugenol with Silyl Ethers:

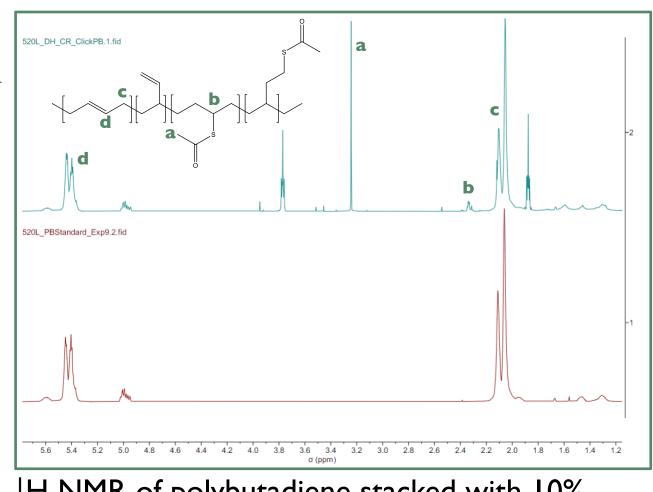




Functionalization of polybutadiene:

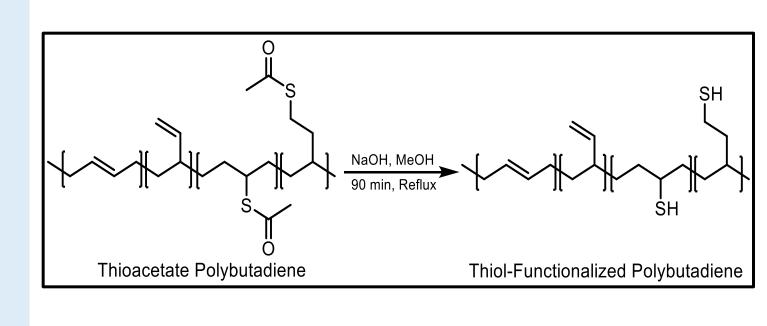


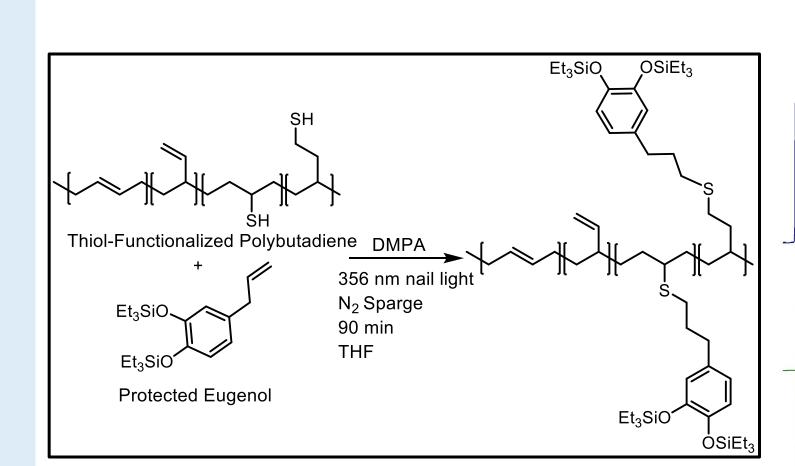
IR Spectrum of 10% functionalized polybutadiene



¹H NMR of polybutadiene stacked with 10% functionalized polybutadiene

Polymer Deprotection & Thiol-ene Reaction





- Thiol-ene reaction was unsuccessful
- Orange circles indicate protected eugenol alkene
- Alkene resonance should disappear from starting material to product, but doesn't



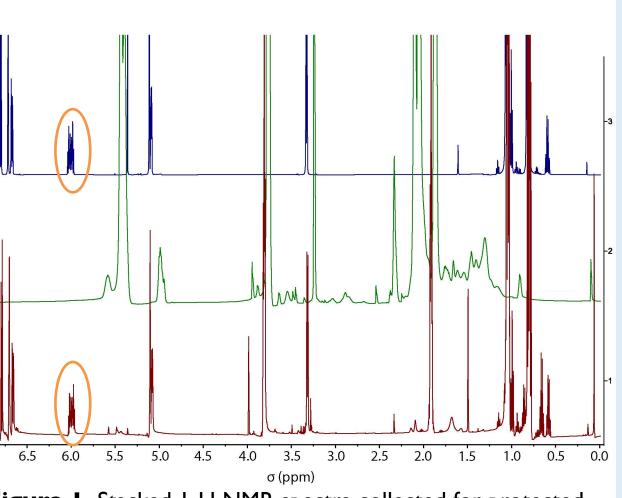
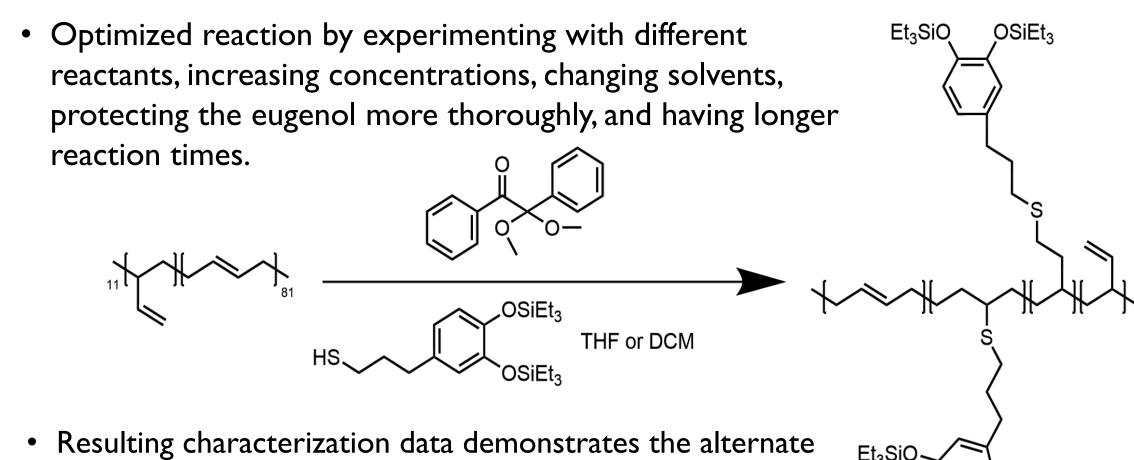
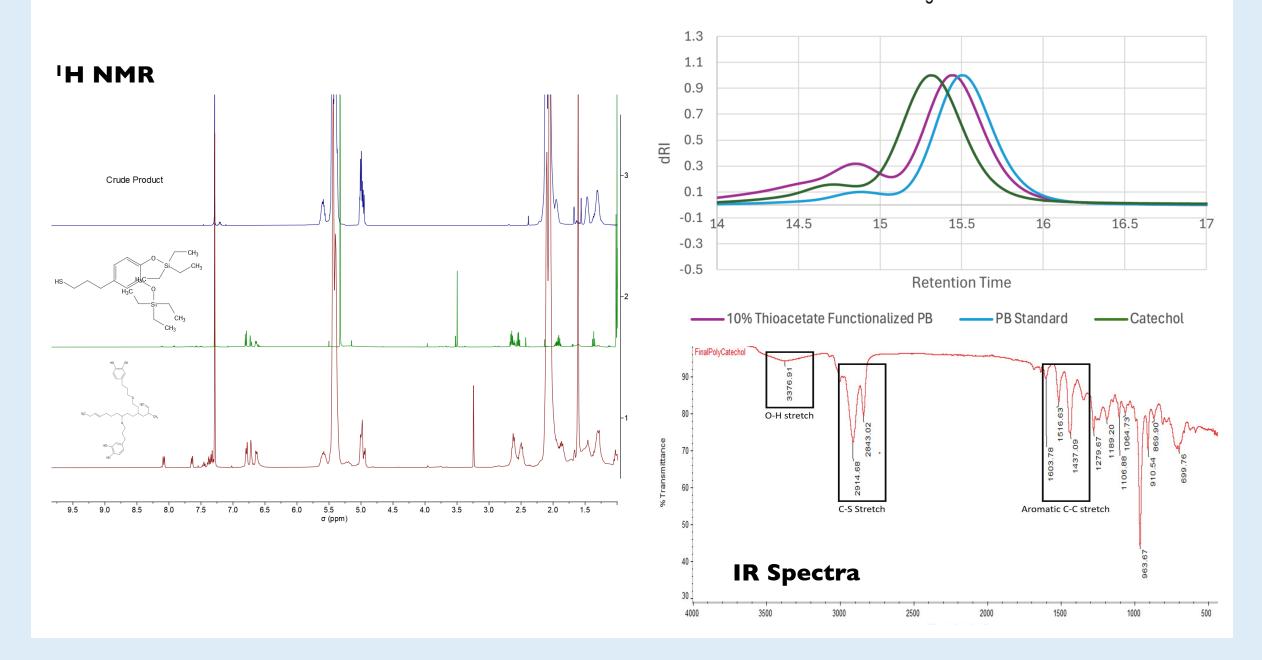


Figure 1. Stacked 1-H NMR spectra collected for protected eugenol (top), polybutadiene (middle), and the thiol-ene reaction product (bottom).

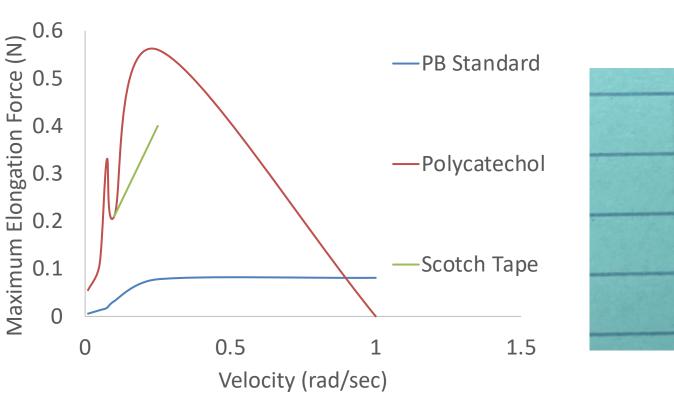
Alternative Route to Catechol Polymer



route was successful in producing desired product.

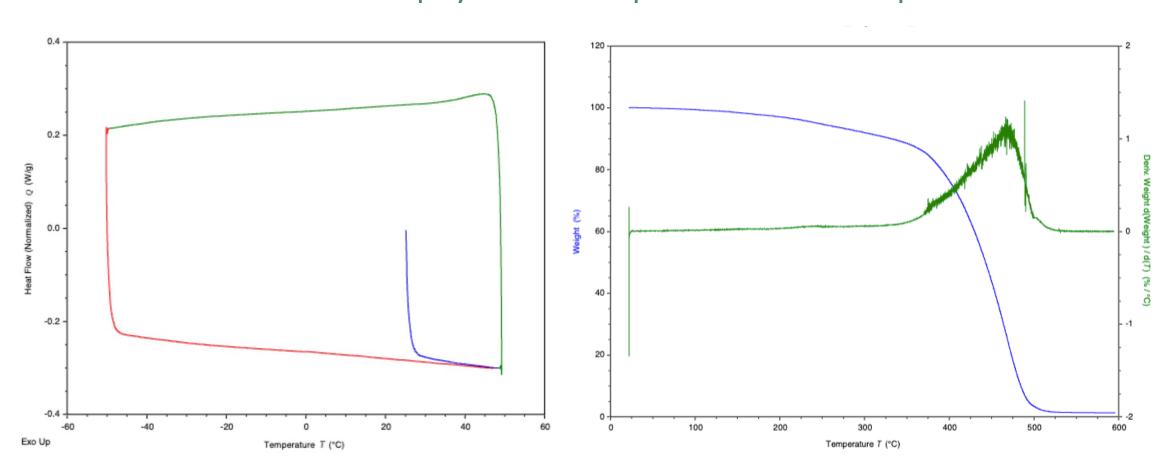


Polymer Properties & Adhesion Testing



1.46 cm • Adhesion forces at a variety of velocities (peel-testing) on polyethylene rectangles were measured. Our novel polycatechol outperformed scotch tape at 0.25 rad/sec.

2.54 cm



• Differential scanning calorimetry (DSC) indicates the polycatechol has a glass transition temperature (T_g) of -47.5 °C.

Thermal gravimetric analysis (TGA) suggests that operating use of the polycatechol should not exceed 50 °C (when degradation begins).